

Ion charge storage in supercapacitor nanopores quantified by modeling and *in situ* SAXS

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Tracking the structural arrangement of ions in carbon supercapacitor nanopores using *in situ* small-angle X-ray scattering†

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A carbon nanopore model to quantify structure and kinetics of ion electrosorption with *in situ* small-angle X-ray scattering†

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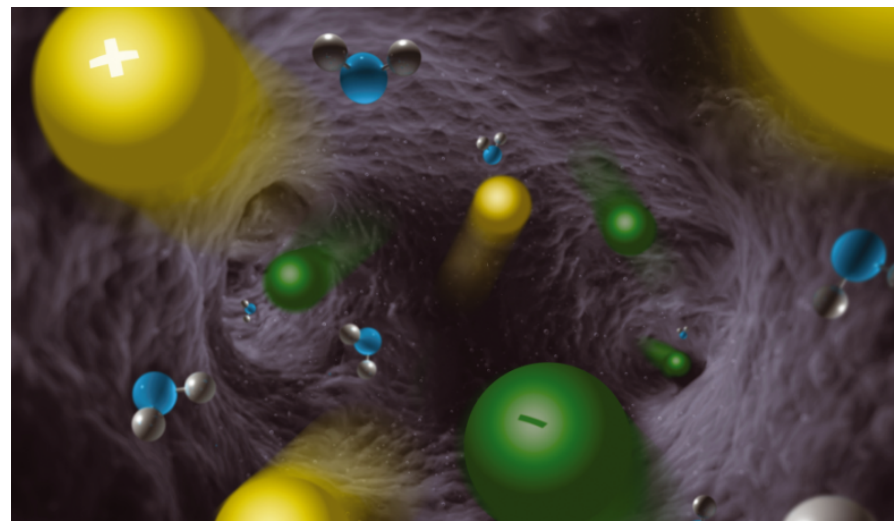
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ARTICLES

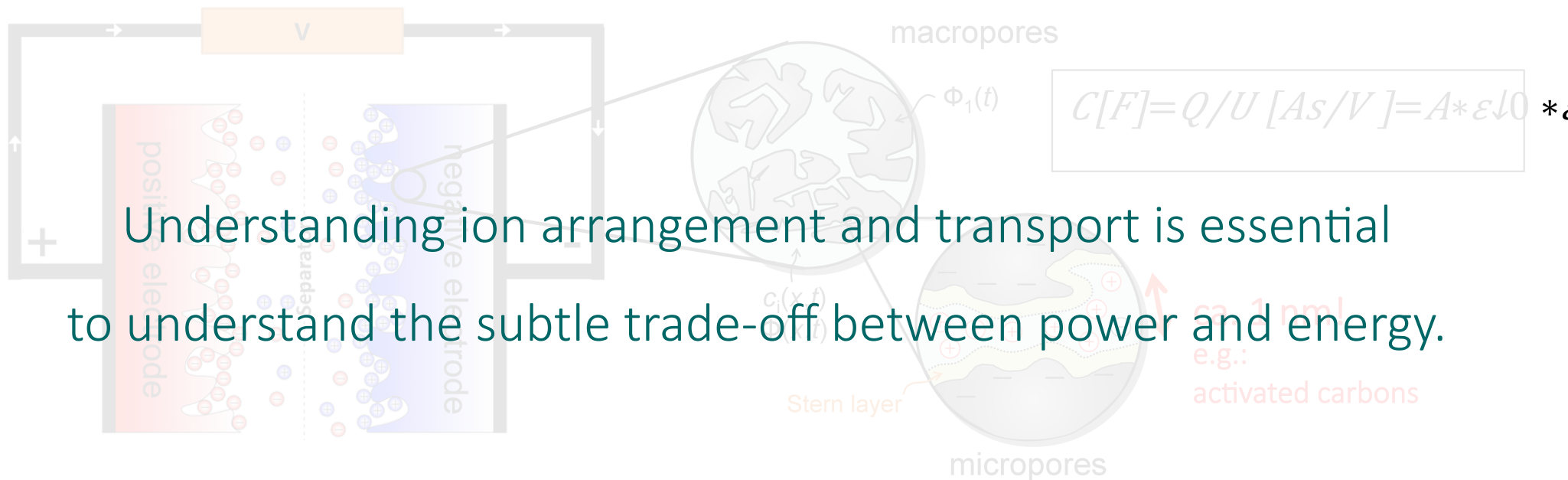
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Quantification of ion confinement and desolvation in nanoporous carbon supercapacitors with modelling and *in situ* X-ray scattering

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- ▶ Effective energy storage devices with high power densities.
- ▶ We study activated and carbide derived carbons with aqueous 1M CsCl
- ▶ Making pores smaller: energy density \uparrow but power density \downarrow



Adapted from F. Béguin, V. Presser, A. Balducci and E. Frackowiak, *Advanced Materials*, 2014, **26**, 2219-2251.

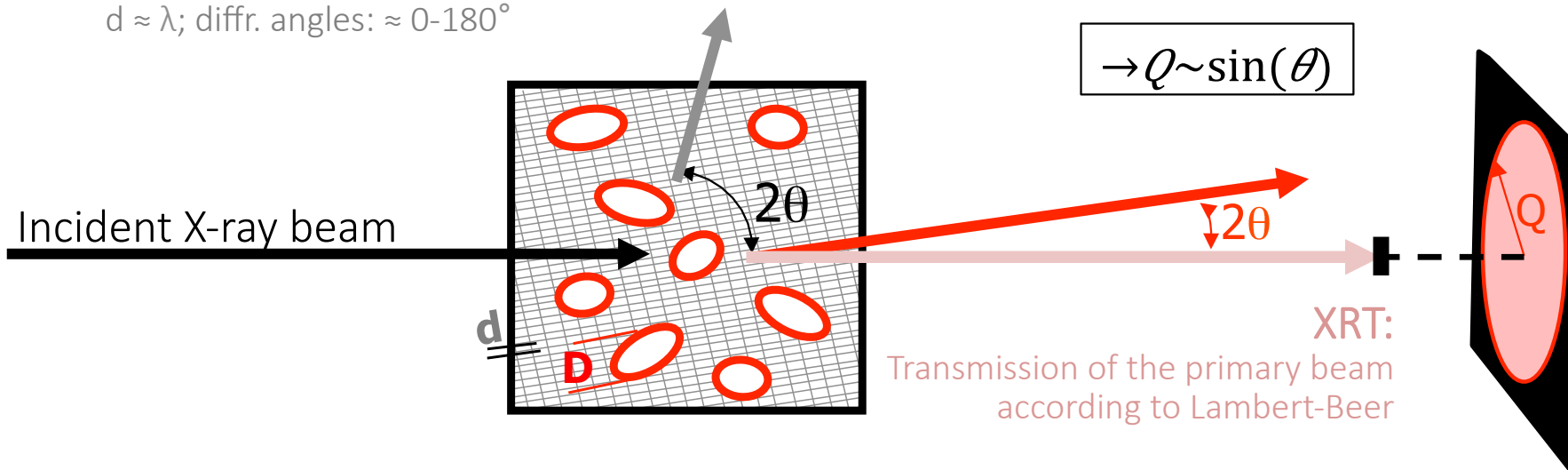
X-ray scattering experiment at Synchrotron:

WAXS:

diffraction (peaks)/scattering from atoms or molecules on crystal lattice/disordered structure;
 $d \approx \lambda$; diffr. angles: $\approx 0-180^\circ$

SAXS:

diffuse scattering from particles, pores, ...
 (electron density fluctuations);
 $D \gg \lambda$; scattering angles: $\approx 0-10^\circ$

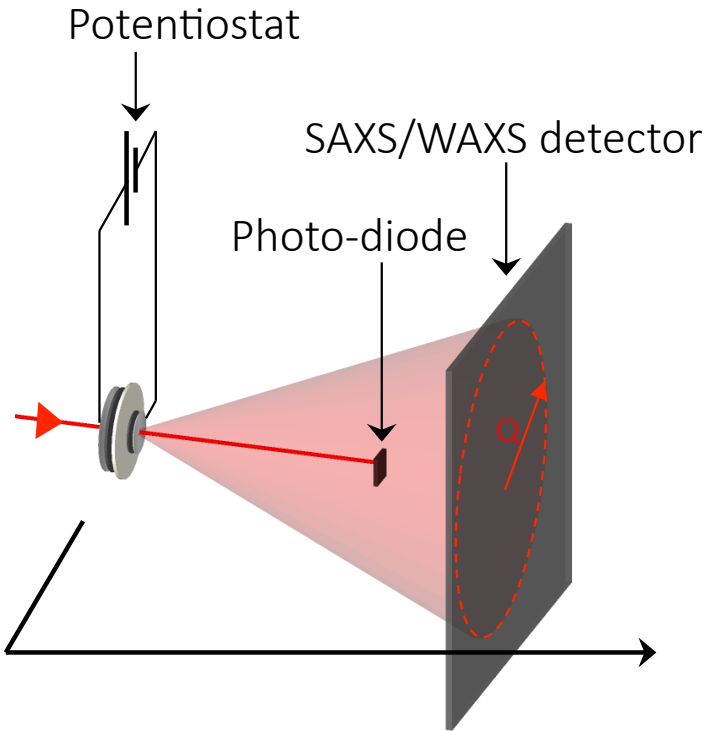


Why using a Synchrotron Radiation Source?

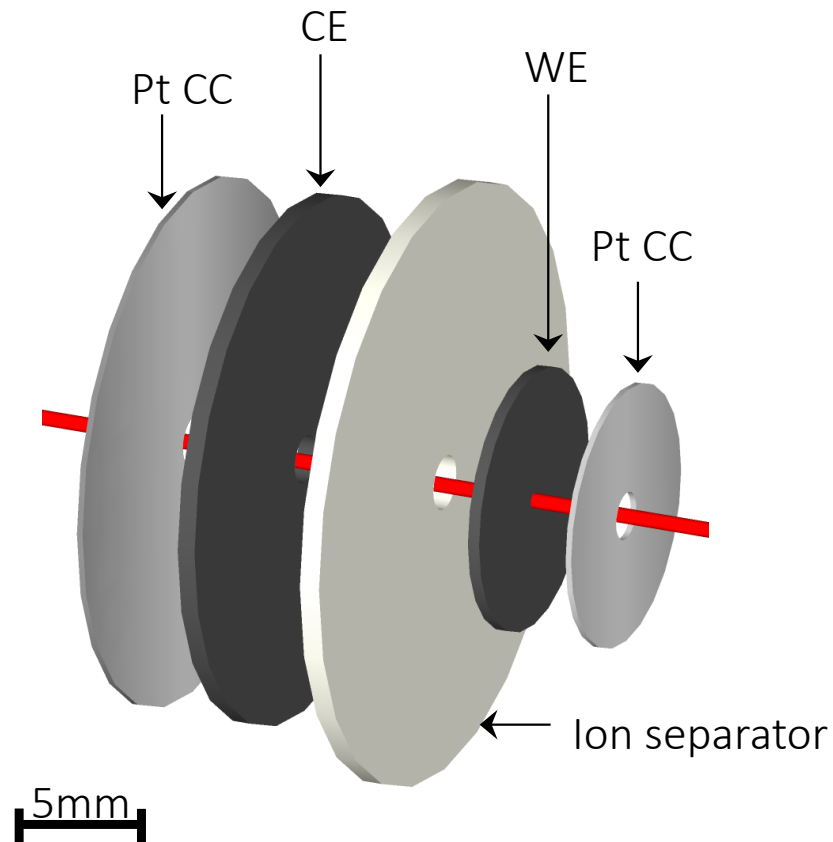
- ▶ Provides extremely high photon flux
- ▶ Time resolutions below one second easily achievable (compared to Lab: $\approx 30\text{min}$)



Elettra Sincrotrone Trieste



20cm



5mm

In-situ cells



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- ▶ Austrian SAXS Beamline (TU Graz) at ELETTRA

In situ supercapacitor cell

Vacuum chamber

WAXS detector

SAXS detector

Biggest advantages:

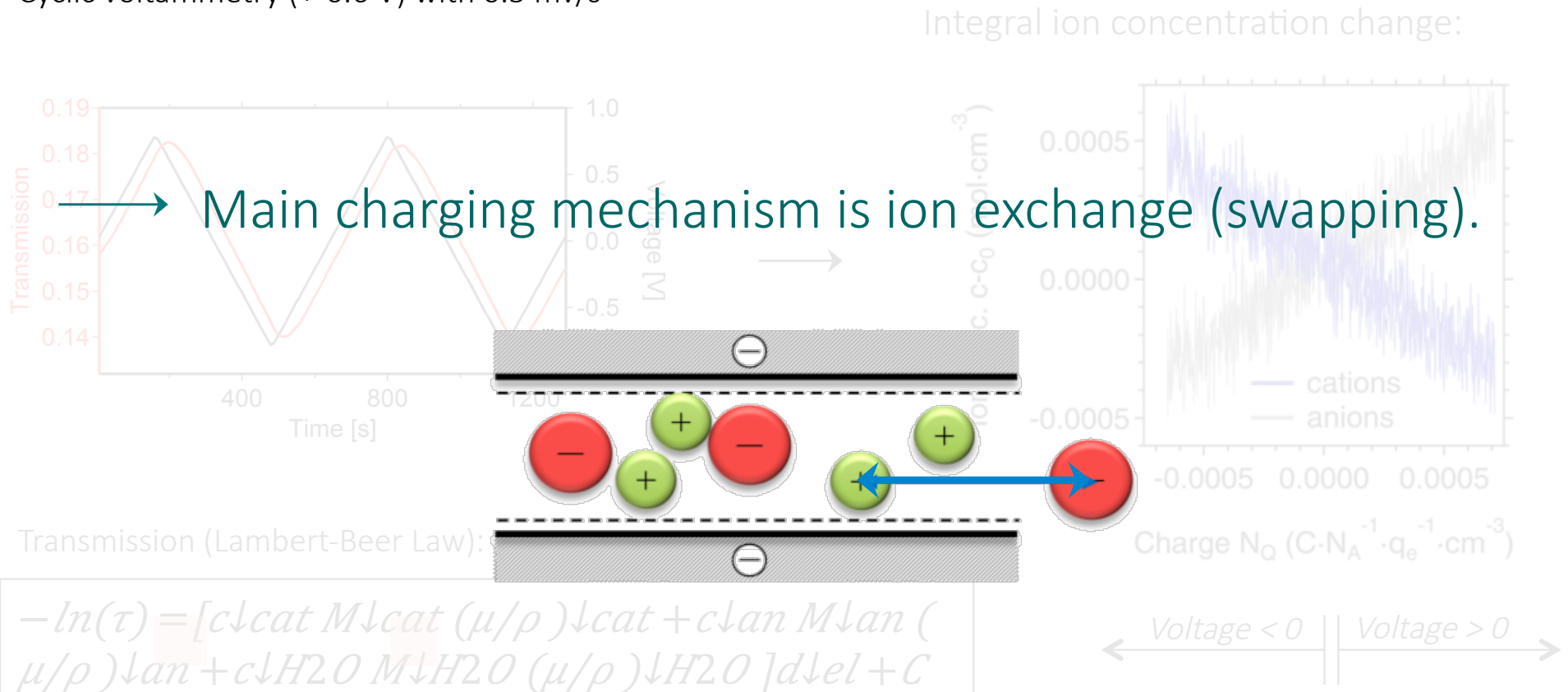
- ▶ Flexibility in terms of set-up
- ▶ Optimized for in situ studies
- ▶ Time resolutions of milliseconds easily achievable
(e.g. pump probe set-up for supercapacitors)

Sample stage

Potentiostat

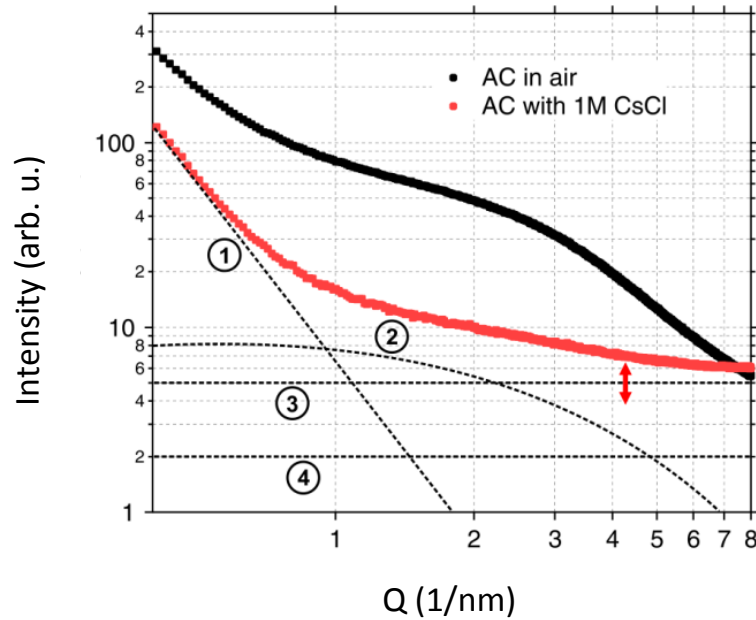
X-ray transmission: counting the ion flux

Here: AC2 (pore size 0.9 nm) with 1M CsCl;
Cyclic voltammetry (+-0.6 V) with 0.5 mv/s



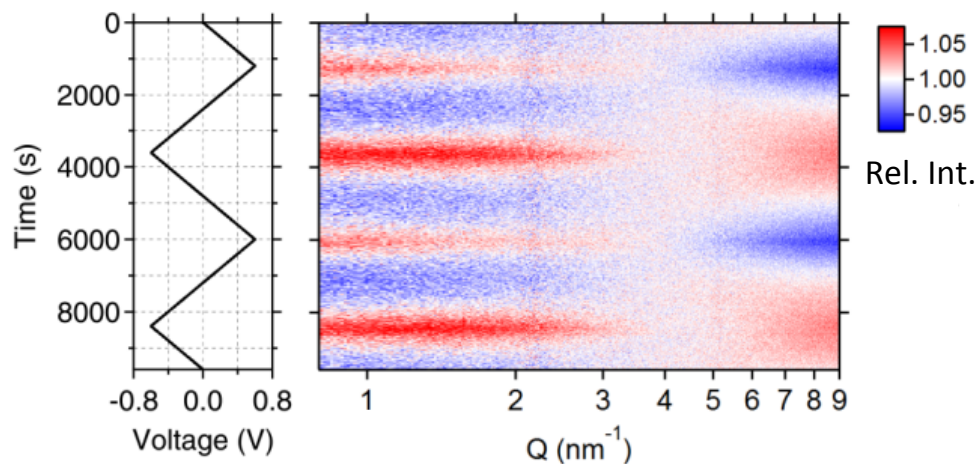
Main charging mechanism is ion exchange (swapping).

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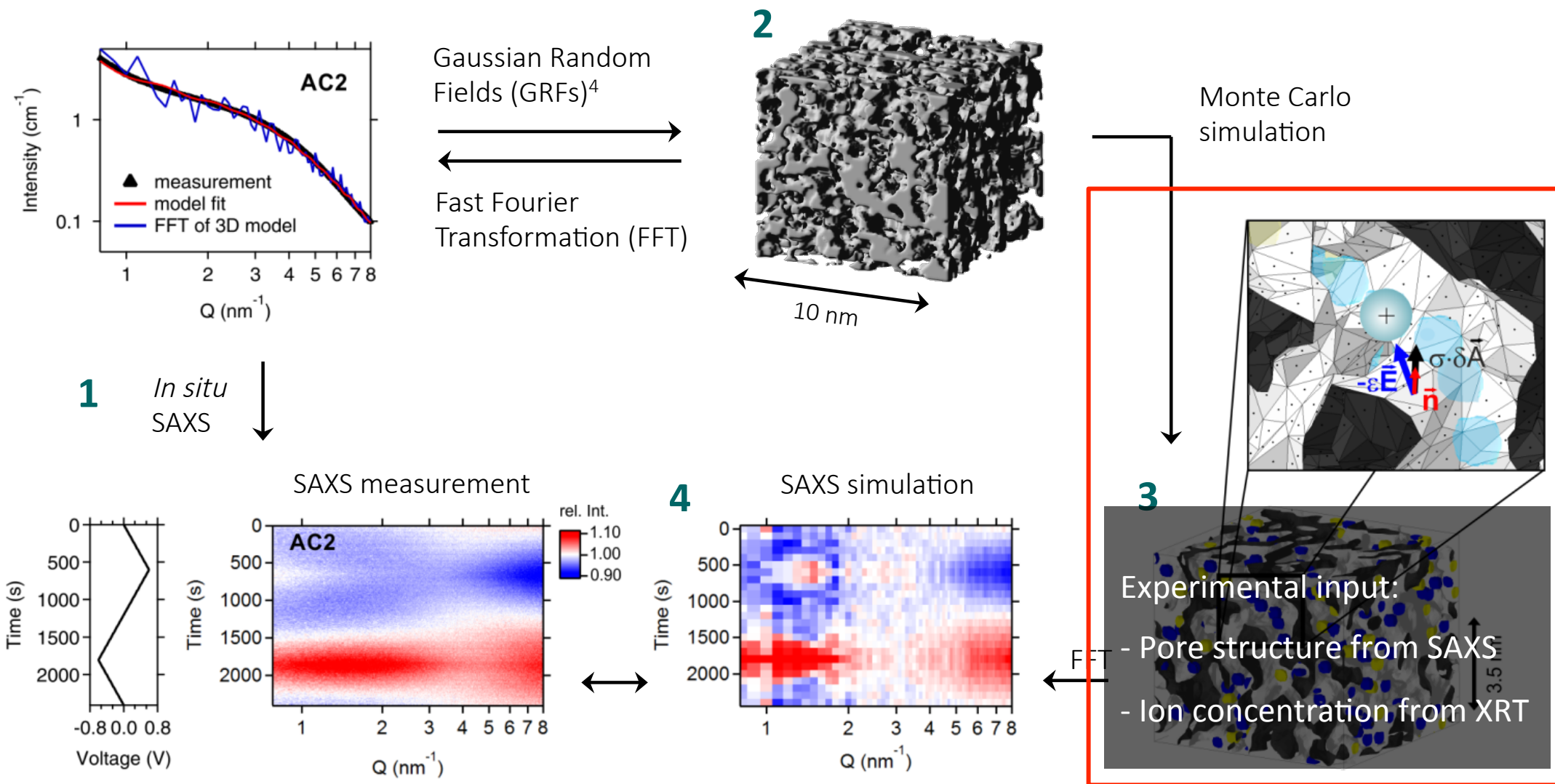


- ① Porod contribution
- ② SAXS contribution (nanopores)
- ③ Electrolyte structure factor
- ④ Carbon structure factor

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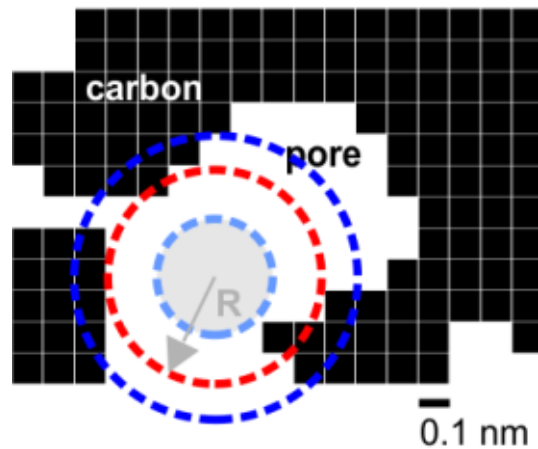
- Rich on information, but interpretation difficult
- We see ion concentration changes.
- And collective structural rearrangement of ions.
- Comprehensive data analysis approach necessary



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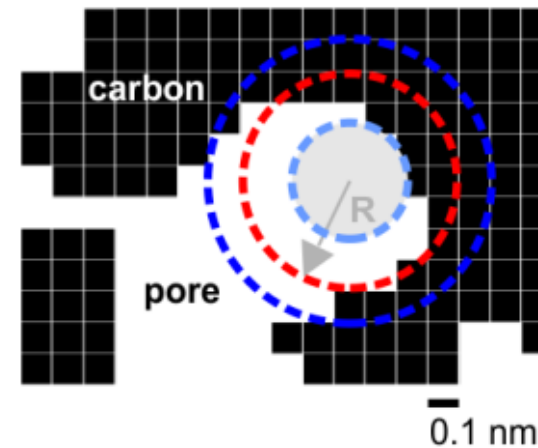
The degree of confinement (DoC) and the degree of desolvation (DoDS)

Small DoC, DoDS



Apply voltage

Large DoC, DoDS



R – bare ion
R – hydrated ion
R – cut-off

$$DoC = \frac{\sum_i \frac{1}{R_i}}{\sum_i \frac{1}{R_i} + C} \quad \text{for } R_i < R_{HS}$$

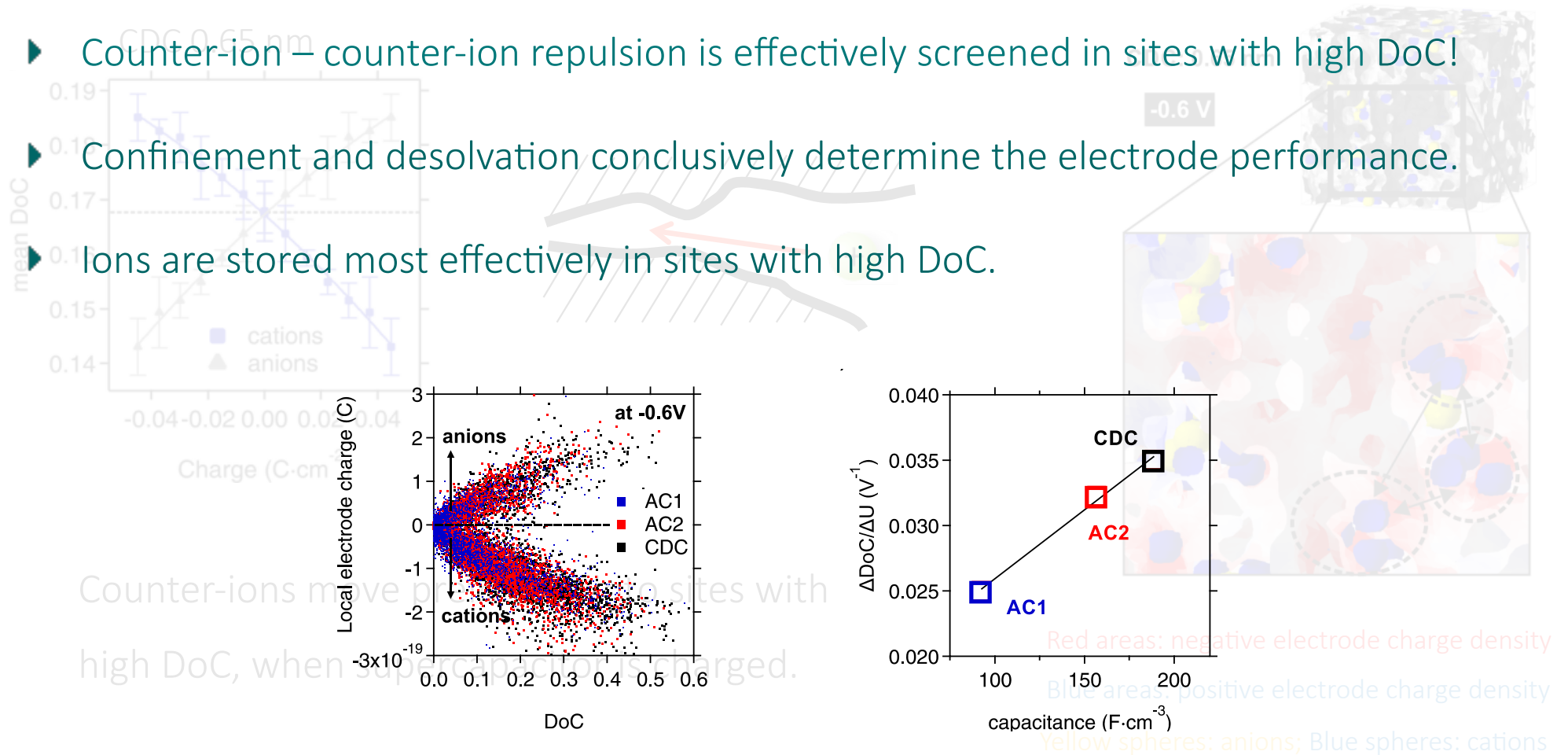
$$DoDS = \frac{\sum_i \frac{1}{R_i} - \sum_i \frac{1}{R_{HS}}}{\sum_i \frac{1}{R_i}} \quad \text{for } R_i < R_{HS}$$

- ▶ The DoC precisely accounts for the ion position in a disordered pore structure.
- ▶ The DoDS is proportional to the number of released water molecules within the hydration shell.

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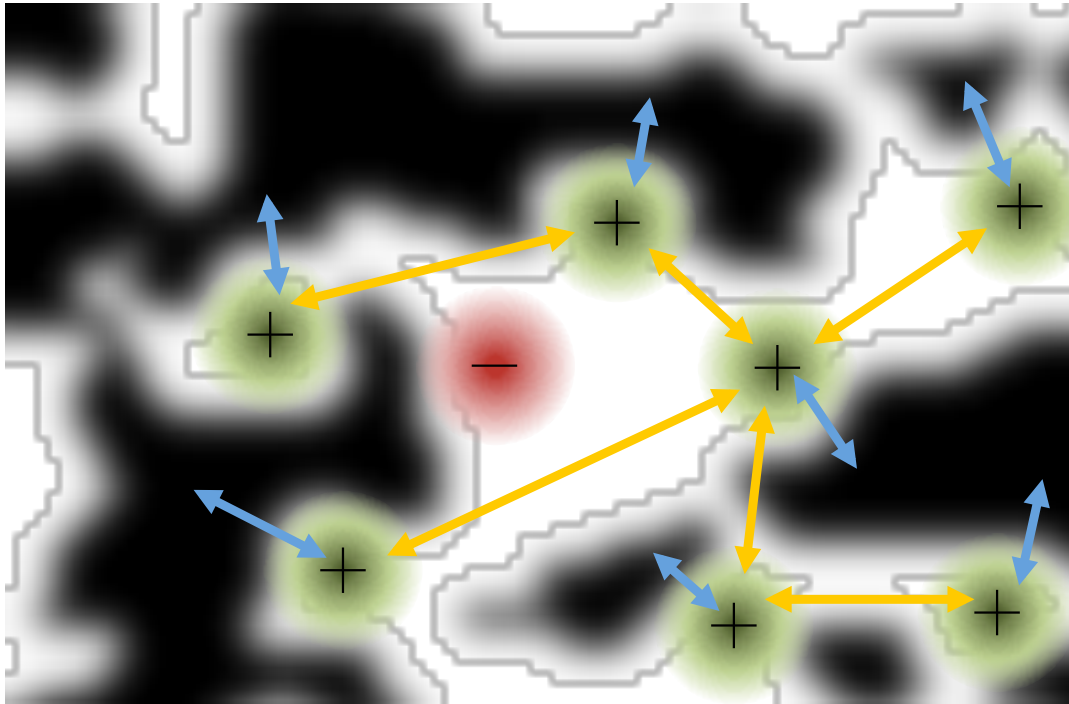
As a voltage is applied ions rearrange on a local scale, i.e. they change their DoC

- ▶ Counter-ion – counter-ion repulsion is effectively screened in sites with high DoC!
- ▶ Confinement and desolvation conclusively determine the electrode performance.
- ▶ Ions are stored most effectively in sites with high DoC.



C. Prehal, C. Koczwara, N. Jäckel, A. Schreiber, M. Burian, H. Amenitsch, M. A. Hartmann, V. Presser and O. Paris, *Nat. Energy*, 2017, **2**, 16215.

Electrode charge = $6 e^-$



- 1) Global concentration change: XRT
- 2) Local ion rearrangement: SAXS
 - i. to minimize repuls. interact. of counter-ions
→ minimized in sites with high DoC
 - ii. Ion-ion correlations remain constant
 - iii. If number of sites with high DoC \uparrow
→ counter-ion density \uparrow
→ capacitance \uparrow

EDLC performance: we need pore structures, providing the maximum amount of pore sites with high DoC.

Thank you for your attention!

